

Tetrahedron Letters 44 (2003) 7649-7653

Stereoselective synthesis of the α -glucosidase inhibitor nectrisine

Alison N. Hulme* and Charles H. Montgomery

School of Chemistry, The University of Edinburgh, West Mains Road, Edinburgh EH9 3JJ, UK Received 19 June 2003; revised 30 July 2003; accepted 8 August 2003

Abstract—The α -glucosidase inhibitor nectrisine was synthesised in 12 steps (31% overall yield) starting from D-serine. The three contiguous stereocentres of this iminosugar were introduced via a highly diastereoselective boron mediated glycolate aldol reaction.

© 2003 Elsevier Ltd. All rights reserved.

1. Introduction

Oligosaccharides have been shown to play important roles in a large number of biological recognition events that range from cell-cell communication, fertilization, and cell differentiation, to pathological processes including cancer metastasis, and inflammation. 1 Iminoanalogues of sugars (also known as azasugars) are known to behave as glycosidase² or glycosyltransferase³ inhibitors by acting as transition state mimics, where the flattened pyrrolidine ring and protonated nitrogen mimic the distorted structure of the glycosyl cation intermediate.⁴ Hence, this class of sugar mimetics offers therapeutic potential in areas such as inflammatory diseases, lysosomal storage disorders, and cancer.⁵ Two such iminosugars are the pyrrolidine 1,4-dideoxyimino-D-arabinitol (DAB-1) 1⁶ and the polyhydroxylated dihydropyrrole nectrisine (FR 900483) 2⁷ (Fig. 1). These compounds are extremely potent α-glucosidase inhibitors $[IC_{50} \ 1.8 \times 10^{-7} \ and \ 4.8 \times 10^{-8} \ M$, respectively (yeast α-glucosidase)],⁸ and the synthesis of nectrisine 2 and related unsaturated analogues has attracted the attention of a number of groups.9 We recently reported an extremely efficient synthesis of DAB-1 (1) from serine-derived aldehyde 3.¹⁰ In this paper, we further

demonstrate the utility of aldehyde 3 by disclosing our synthesis of the related iminosugar nectrisine 2.

2. Results and discussion

A number of serine derived aldehydes can be found in the literature. 11 Unfortunately, many of these demonstrate a propensity to undergo racemisation and/or exhibit poor diastereoselectivity under nucleophilic addition. 12 However, the addition of nucleophiles to N,N-dibenzyl protected amino aldehydes has been shown to take place with high diastereoselectivity under both chelation and non-chelation controlled conditions making aldehyde 3 an attractive building block.¹³ The key step in our synthetic strategy was the construction of the diol motif in 4 which we envisaged could be achieved via either dihydroxylation of a γ -amino- α , β unsaturated ester, or a boron-mediated asymmetric aldol condensation. Protecting group 'tuning' of the dihydroxylation reaction of γ -amino- α , β -unsaturated ester derivatives has been reported for a range of less complex amino acid derivatives,14 and the dihydroxylation reaction of a number of carbamate and oxazoline protected serine derivatives has also been reported.¹⁵

Figure 1. Retrosynthetic analysis of the polyhydroxylated plant alkaloids DAB-1 (1) and nectrisine 2.

Keywords: iminosugars; glycolate aldol; glycosidase inhibitors.

^{*} Corresponding author. Tel.: +44-(0)131-650-4711; fax: +44-(0)131-650-4743; e-mail: alison.hulme@ed.ac.uk

The dihydroxylation reaction of γ -amino- α , β -unsaturated ester derivatives has recently been used in the synthesis of several natural product targets including deoxyaminohexoses, ^{16a} polyhydroxylated indolizidine plant alkaloids, ^{16b,c} and α , β -dihydroxy- γ -aminoacids ^{16d} including hydroxystatine isosteres, ^{16e} and the AI-77's. ^{16f} On the other hand, the glycolate aldol reaction has recently been shown to represent a viable alternative to the dihydroxylation reaction, offering both *syn* and *anti* diastereocontrol. ¹⁷

The synthesis of aldehyde 3 was achieved in five steps from D-serine with 88% overall yield and >98% ee, using our previously reported procedure. 10,18 This aldehyde was found to be stable to racemisation on storage at 4°C for up to 48 h. Aldehyde 3 was converted to the α,β-unsaturated ester 5 in excellent yield (95%) under Horner-Wadsworth-Emmons conditions using the mild base Ba(OH)₂ (Scheme 1).¹⁹ Osmium tetroxide catalysed dihydroxylation resulted in a moderate yield (59%) of diols 6 and 7 in a 65:35 diastereomeric ratio, favouring the undesired all-syn diol 6.20,21 The use of AD-mix-α resulted in a very sluggish reaction and only a modest improvement in yield (65% based on unrecovered starting material) and no apparent increase in diastereoselectivity. However, the diastereoselectivity could be overturned in favour of the desired syn diol 7 by use of AD-mix- β , but again the reaction rate was very slow (7 days) and the diastereoselectivity modest (32:68, 6:7). Separation of the desired *syn* diastereomer 7, conversion to the corresponding Weinreb amide and removal of the *N*-benzyl protecting groups using Pearlman's catalyst resulted in conversion in situ to the previously reported lactam $8.^{10}$ In an attempt to improve the alkene reactivity towards dihydroxylation, ester 5 was reduced to alcohol 9. This allylic alcohol was found to undergo dihydroxylation more readily than its ester counterpart (75% yield, 60 h), but similarly disappointing levels of diastereoselectivity (\sim 3:1) were observed, deterring further investigation of this route.

We have previously shown that reaction of aldehyde 3 with Evan's chiral glycolate equivalent 10^{22} affords the desired *syn* aldol adduct (Felkin–Anh product) 11 in excellent yield as a single diastereomer (Scheme 2). Aldol adduct 11 may be converted to pyrrolidinone 8 via Weinreb amide 12 in 71% over the two steps. This combination of Evans oxazolidinone and α -chiral aldehyde was shown to represent a 'matched' stereo-inductive pairing, by the stereochemical outcome of the reaction of methyl benzyloxyacetic acid 13^{24} with aldehyde 3 which also provided a single diastereomeric aldol adduct 14^{25} in good yield (75%). The identity of

TBDPSO
$$\stackrel{\bullet}{\underset{NB}{\bigvee}}$$
 H $\stackrel{\bullet}{\underset{NB}{\longrightarrow}}$ TBDPSO $\stackrel{\bullet}{\underset{NB}{\bigvee}}$ OMe $\stackrel{\bullet}{\underset{NB}{\longrightarrow}}$ TBDPSO $\stackrel{\bullet}{\underset{NB}{\bigvee}}$ OH $\stackrel{\bullet}{\underset{NB}{\longrightarrow}}$ TBDPSO $\stackrel{\bullet}{\underset{NB}{\longrightarrow}}$ OH $\stackrel{\bullet}{\underset{NB}{\longrightarrow}}$ TBDPSO $\stackrel{\bullet}{\underset{NB}{\longrightarrow}}$ OH $\stackrel{\bullet}{\underset{NB}{\longrightarrow}}$ TBDPSO $\stackrel{\bullet}{\underset{NB}{\longrightarrow}}$ OH $\stackrel{\bullet}{\underset{NB}{\longrightarrow}}$

Scheme 1. Reagents and conditions: (a) (MeO)₂POCH₂CO₂Me, Ba(OH)₂ (activated), THF:H₂O (40:1), rt, 1 h; **3**, THF, rt, 18 h (95%); (b) OsO₄ (18 mol%), NMO, acetone:H₂O (8:1), rt, 5 h (59%, 65:35 ratio **6:7**); (c) AD-mix-α, MeSO₂NH₂, acetone:H₂O (3:1), rt, 7 d (68%, 32:68 ratio **6:7**); (e) LiAlH₄, Et₂O, -78°C, 45 min (85%).

Scheme 2. Reagents and conditions: (a) i. Et₃N, n-Bu₂BOTf, CH₂Cl₂, -78°C to 0°C, 3 h; ii. 3, -78°C to 0°C, 2.5 h (82% for 11, 87% for 17); (b) (MeO)NHMe·HCl, Me₃Al, THF, -30°C; 11 or 17, 0°C, 2.5 h (100% for 12, 90% for 18); (c) i. Pr₂NEt, n-Bu₂BOTf, Et₂O, -78°C, 1.5 h; ii. 3, -78°C to 0°C, 2 h (75%); (d) (MeO)NHMe·HCl, Me₃Al, THF, -30°C; 14, -30°C to rt, 18 h (91%); (e) Pd(OH)₂, H₂, MeOH, rt, 72 h (71% for 8, 65% for 15).

14 was confirmed by its one-step conversion to the previously reported Weinreb amide 12. The same reaction sequence has also allowed the synthesis of the methyl pyrrolidinone analogue $(R = Me, 15)^{26}$ via reaction of acylated Evans auxiliary 16 to give a single diastereomer of aldol adduct 17, conversion of this adduct to Weinreb amide 18 and subsequent deprotection and cyclisation (51% for three steps from 16 to 15).

Aldol addition of the enolate of the enantiomeric glycolate equivalent (ent-10) to aldehyde 3 resulted in the formation of a separable 9:1 diastereomeric mixture of syn aldol adducts 19 and 20 (Scheme 3).²⁷ Thus whilst the Felkin–Anh selectivity of aldehyde 3 is sufficiently high to provide a single diastereomeric adduct in the 'matched' case, or in the absence of a chiral auxiliary, the directing power of aldehyde 3 can easily be overcome under 'mis-matched' conditions. This would allow the generation of a range of aldol stereoisomers with high diastereoselectivities, simply by judicious choice of chiral auxiliary and chirality of aldehyde 3 (derivable from L- or D-serine). Aldol adducts 19 and 20 were converted to Weinreb amides 21 and 12, respectively, on treatment with N,O-dimethylhydroxylamine·HCl

and AlMe₃. All-syn amide **21** was converted to the diastereomeric pyrrolidinone **22** in good yield (69%). This material was found to be identical to that generated by the two-step conversion of all-syn diol **6** to **22**. 28

A number of different routes to the synthesis of nectrisine 2 from either the Weinreb amide 12, or pyrrolidinone 8, were pursued, but we found that partial reduction of the pyrrolidinone followed by dehydration proved the most fruitful (Scheme 4). Direct reduction of **8** with DIBAL-H or Super Hydride[®] gave no reaction. We therefore surmised that introduction of an electron withdrawing protecting group on the nitrogen atom, as has been used in the reduction of other lactams, 9c,29 might improve the yield of the desired pyrrolidinol. Initial attempts to effect a global Boc protection resulted only in the isolation of a compound which was tentatively assigned as dihydropyrrol-2-one 23. Thus prior protection of the diol was required, followed by N-Boc activation to give 24.30 As predicted, the increased carbonyl electrophilicity resulting from N-Boc protection facilitated the smooth reduction of the lactam with Super Hydride[®] even at −78°C to give 25. This strategy was similarly successful for the synthesis

Scheme 3. Reagents and conditions: (a) i. Et₃N, n-Bu₂BOTf, CH₂Cl₂, -78°C to 0°C, 3 h; ii. 3, -78°C to 0°C, 2.5 h (79%, 9:1 ratio **19:20**); (b) (MeO)NHMe·HCl, Me₃Al, THF, -30°C; **19** or **20**, -30°C to 0°C, 24 h (94% for **21**, 93% for **12**); (c) Pd(OH)₂, H₂, MeOH, rt, 72 h (69%); (d) (MeO)NHMe·HCl, Me₃Al, THF, -30°C; **6**, -30°C to rt, 24 h (93%); (e) Pd(OH)₂, H₂, MeOH, rt, 24 h (85%).

Scheme 4. Reagents and conditions: (a) Boc₂O, Et₃N, DMAP (cat.), CH₂Cl₂, rt, 30 min (86%); (b) TBSOTf, 2,6-lutidine, CH₂Cl₂, rt, 3 h (85% for R = OTBS, 94% for R = Me); (c) Boc₂O, Et₃N, DMAP (cat.), CH₂Cl₂, rt, 1.5 h (96% for **24**, 97% for **27**); (d) LiEt₃BH, THF, -78°C, 15 min (96% for **25**, 93% for **26**); (e) i. 6N HCl (aq.), THF, 50°C, 2 h; ii. Dowex 1X2 (HO⁻), H₂O (80%).

of the methyl analogue **26** via Boc-activated pyrrolidinone **27** (85% yield over three steps from **15**). Heating amino alcohol **25** with 6N HCl at 50°C for 2 h led to the clean removal of all of the protecting groups and the formation of an intermediate aminosugar **28**. See Neutralisation and purification by ion-exchange chromatography [Dowex 1X2 (HO⁻)] provided nectrisine **2** in excellent yield.

In conclusion, we have synthesized nectrisine 2 in 12 steps from D-serine with an overall yield of 32%. The key carbon—carbon bond forming reaction, a glycolate aldol condensation, was shown to take place with excellent diastereoselectivity. Furthermore, the aldol based approach offers a flexible strategy for the synthesis of a range of stereoisomers and other analogues of this iminosugar.

Acknowledgements

The authors gratefully acknowledge the EPSRC (GR/96309957 to CHM) for funding this work.

References

- (a) Bertozzi, C. R.; Kiessling, L. L. Science 2001, 291, 2357–2364; (b) Yarema, K. J.; Bertozzi, C. R. Curr. Opin. Chem. Biol. 1998, 2, 49–61; (c) Dwek, R. A. Chem. Rev. 1996, 96, 683–720.
- 2. Legler, G. In *Iminosugars as Glycosidase Inhibitors*; Stütz, A. E., Ed.; Wiley-VCH: New York, 1998; Chapter 3.
- Compain, P.; Martin, O. R. Curr. Top. Med. Chem. 2003, 3, 541–560.
- McGarvey, G. J.; Wong, C.-H. Liebigs Ann./Rec. 1997, 1059–1074.
- (a) Asano, N. Curr. Top. Med. Chem. 2003, 3, 471–484;
 (b) Butters, T. D.; Dwek, R. A.; Platt, F. M. Chem. Rev. 2000, 100, 4683–4696;
 (c) Nangia-Makker, P.; Conklin, J.; Hogan, V.; Raz, A. Trends Mol. Med. 2002, 8, 187–192.
- (a) Furukawa, J.; Okuda, S.; Saito, K.; Hatanaka, S. I. *Phytochemistry* 1985, 24, 593–594; (b) Nash, R. J.; Bell, E. A.; Williams, J. M. *Phytochemistry* 1985, 24, 1620–1622.
- 7. (a) Shibata, T.; Nakayama, O.; Tsurumi, Y.; Okuhara, M.; Terano, H.; Kohsaka, M. J. Antibiot. 1988, 41, 296–301; (b) Kayakiri, H.; Takase, S.; Setoi, H.; Uchida, I.; Terano, H.; Hashimoto, M. Tetrahedron Lett. 1988, 29, 1725–1728.
- (a) DAB-1: Fleet, G. W. J.; Nicholas, S. J.; Smith, P. W.; Evans, S. V.; Fellows, L. E.; Nash, R. J. *Tetrahedron Lett.* 1985, 26, 3127–3130; (b) nectrisine: Tsujii, E.; Muroi, M.; Shiragami, N.; Takatsuki, A. *Biochem. Biophys. Res. Commun.* 1996, 220, 459–466.
- (a) Tschamber, T.; Siendt, H.; Tarnus, C.; Deredas, D.; Frankowski, A.; Kohler, S.; Streith, J. Eur. J. Org. Chem. 2002, 702–712; (b) Bosco, M.; Bisseret, P.; Bouix-Peter, C.; Eustache, J. Tetrahedron Lett. 2001, 42, 7949–7952; (c) Kim, Y. J.; Takatsuki, A.; Kogoshi, N.; Kitahara, T. Tetrahedron 1999, 55, 8353–8364; (d) Takayam, S.; Mar-

- tin, R.; Wu, J. Y.; Laslo, K.; Siuzdak, G.; Wong, C. H. J. Am. Chem. Soc. 1997, 119, 8146–8151; (e) Kayakiri, H.; Nakamura, K.; Takase, S.; Setoi, H.; Uchida, I.; Terano, H.; Hashimoto, M.; Tada, T.; Koda, S. Chem. Pharm. Bull. 1991, 39, 2807–2812; (f) Chen, S.-H.; Danishefsky, S. J. Tetrahedron Lett. 1990, 31, 2229–2232.
- Hulme, A. N.; Montgomery, C. H.; Henderson, D. K. J. Chem. Soc., Perkin Trans. 1 2000, 1837–1841.
- (a) Fehrentz, J. A.; Castro, B. Synthesis 1983, 676–678;
 (b) Garner, P.; Park, J. M. J. Org. Chem. 1987, 52, 2361–2364;
 (c) Herborn, C.; Zietlow, A.; Steckhan, E. Angew. Chem., Int. Ed. Engl. 1989, 28, 1399–1401;
 (d) Lubell, W.; Rappaport, H. J. Org. Chem. 1989, 54, 3824–3831;
 (e) Reetz, M. T. Angew. Chem., Int. Ed. Engl. 1991, 30, 12.
- Jurzac, J.; Golebiowski, A. Chem. Rev. 1989, 89, 149– 164.
- (a) Reetz, M. T. Chem. Rev. 1999, 99, 1121–1162; (b) Reetz, M. T. Angew. Chem., Int. Ed. Engl. 1991, 30, 1531–1750.
- Reetz, M. T.; Strack, T. J.; Mutulis, F.; Goddard, R. Tetrahedron Lett. 1996, 37, 9293–9296.
- (a) Ikota, N.; Hirano, J.; Gamage, R.; Nakagawa, H.; Hama-Inaba, H. Heterocycles 1997, 46, 637–644; (b) Huang, Y.; Dalton, D. R.; Carroll, P. J. J. Org. Chem. 1997, 62, 372–376.
- (a) Koskinen, A. M. P.; Otsomaa, L. A. Tetrahedron 1997, 53, 6473-6484; (b) Gurjar, M. K.; Ghosh, L.; Syamala, M.; Jayasree, V. Tetrahedron Lett. 1994, 35, 8871-8872; (c) Kim, N.-S.; Kang, C. H.; Cha, J. K. Tetrahedron Lett. 1994, 35, 3489-3492; (d) Palomo, C.; Oiarbide, M.; Landa, A.; Esnal, A.; Linden, A. J. Org. Chem. 2001, 66, 4180-4186; (e) Dolle, R. E.; Herpin, T. F.; Shimshock, Y. C. Tetrahedron Lett. 2001, 42, 1855-1858; (f) Broady, S. D.; Rexhausen, J. E.; Thomas, E. J. J. Chem. Soc., Perkin Trans. 1 1999, 1083-1094.
- (a) Crimmins, M. T.; McDougall, T. J. Org. Lett. 2003, 5, 591–594; (b) Andrus, A. B.; Meredith, E. L.; Simmons, B. L.; Sekhar, B. B. B. V. S.; Hicken, E. J. Org. Lett. 2002, 4, 3549–3552; (c) Davies, S. G.; Nicholson, R. L.; Smith, A. D. Synlett 2002, 1637–1640; (d) Hulme, A. N.; Rosser, E. M. Org. Lett. 2002, 4, 265–267.
- Hulme, A. N.; Curley, K. S. J. Chem. Soc., Perkin Trans. 1 2002, 1083–1091.
- Paterson, I.; Yeung, K. S.; Smaill, J. B. Synlett 1993, 774–776.
- 20. Compound **6** was unambiguously determined as the all-syn product by single crystal X-ray diffraction.
- 21. All new compounds gave spectroscopic data in agreement with their assigned structures: **6**; mp 112–113°C; $[\alpha]_D$ –32.9 (c 1.4, CHCl₃); v_{max} (neat)/cm⁻¹ 3530, 1747; δ_H (250 MHz, CDCl₃) 7.73–7.17 (20H, m), 4.30 (1H, br s), 3.97–3.87 (4H, m), 3.94 (2H, d, J=13.0), 3.78 (3H, s), 3.53 (2H, d, J=13.0), 3.13 (1H, ddd \sim dt, J=8.2, 4.0), 2.85 (1H, br s), 1.11 (9H, s); δ_C (50.3 MHz, CDCl₃) 173.3, 138.5 (2C), 135.6 (2C), 135.6 (2C), 132.5 (2C), 130.0, 129.9, 129.0 (4C), 128.4 (4C), 127.8 (4C), 127.3 (2C), 70.5, 68.2, 59.5, 59.0, 54.4 (2C), 52.2, 26.7 (3C), 18.9. 7; $[\alpha]_D$ –30.6 (c 1.15, CHCl₃); v_{max} (neat)/cm⁻¹ 3482, 1738; δ_H (250 MHz, CDCl₃) 7.74–7.20 (20H, m), 4.66 (1H, br s), 4.12–4.07 (4H, m), 4.06 (2H, d, J=13.4), 3.77 (3H, s), 3.53 (2H, d, J=13.4), 3.04 (1H, dt, J=9.4, 5.7), 2.75 (1H, br s), 1.08 (9H, s); δ_C (50 MHz, CDCl₃) 174.3,

- 139.3 (2C), 135.6 (2C), 135.5 (2C), 132.6, 132.5, 129.9 (2C), 128.9 (4C), 128.2 (4C), 127.8 (4C), 127.0 (2C), 72.4, 70.5, 61.5, 58.7, 54.9 (2C), 52.3, 26.7 (3C), 18.9.
- 22. Fuhry, M. A. M.; Holmes, A. B.; Marshall, D. R. *J. Chem. Soc.*, *Perkin Trans.* 1 **1993**, 2743–2746.
- Masamune, S.; Choy, W.; Peterson, J. S.; Sita, L. R. Angew. Chem., Int. Ed. Engl. 1985, 24, 1–30.
- 24. Synthesized via reaction of the sodium salt of benzyl alcohol with bromoacetic acid, followed by esterification.
- 25. **14**; $[\alpha]_D$ -32.2 (c 1.94, CHCl₃); $v_{\rm max}$ (neat)/cm⁻¹ 3546, 1751, 1601, 1588, 1494; $\delta_{\rm H}$ (250 MHz, CDCl₃) 7.82–7.15 (25H, m), 4.40 (1H, d, J=10.2), 4.38 (1H, d, J=1.4), 4.18 (1H, dd, J=10.7, 5.6), 4.17–4.10 (1H, m), 4.09 (1H, dd, J=10.7, 5.6), 3.91 (2H, d, J=13.5), 3.77 (3H, s), 3.62 (2H, d, J=13.5), 3.52 (1H, d, J=10.2), 3.18 (1H, dt, J=9.1, 5.6), 2.70 (1H, br d, J=9.2), 1.12 (9H, s); $\delta_{\rm C}$ (62.9 MHz, CDCl₃) 172.3, 140.0 (2C), 137.4, 135.6 (2C), 135.5 (2C), 133.1, 132.9, 129.7, 129.6, 129.2 (4C), 128.2 (4C), 127.9 (4C), 127.6 (4C), 127.5, 126.9 (2C), 77.3, 72.8, 72.1, 61.2, 59.1, 54.8 (2C), 51.8, 26.7 (3C), 18.9.
- 26. **15**; $[\alpha]_D$ –6.4 (*c* 0.95, CHCl₃); ν_{max} (neat)/cm⁻¹ 3385, 3272, 1696; δ_H (250 MHz, CDCl₃) 7.66–7.25 (10H, m), 5.86 (1H, br s), 3.80 (1H, dd, J=10.2, 4.5), 3.72 (1H, m), 3.63 (1H, dd, J=10.2, 6.6), 3.52 (1H, dt, J=6.6, 4.5), 2.82 (1H, br s), 2.38 (1H, dq ~qn, J=7.2), 1.18 (3H, d, J=7.2), 1.05 (9H, s); δ_C (62.9 MHz, CDCl₃) 176.6, 135.4 (4C), 132.6, 129.9 (2C), 127.8 (4C), 76.9, 65.1, 61.2 (2C), 44.9, 16.7 (3C), 19.0, 13.0.
- 27. **19**; $[\alpha]_D$ -29.2 (c 0.9, CHCl₃); v_{max} (neat)/cm⁻¹ 3331, 1773, 1708; $\delta_{\rm H}$ (360 MHz, CDCl₃) 7.81–7.77 (4H, m), 7.53-7.22 (24H, m), 7.09-7.06 (2H, m), 5.22 (1H, d, J=2.4), 4.56–4.52 (1H, m), 4.52 (1H, d, J=11.5), 4.50 (1H, ddt, J=9.8, 6.2, 3.1), 4.27 (1H, d, J=11.5), 4.15– 4.11 (1H, m), 4.13 (1H, dd, J=11.3, 4.4) 4.04 (1H, dd, J=11.3, 8.0), 3.93 (2H, d, J=13.1), 3.86–3.81 (3H, m), 3.56 (1H, td, 8.0, 4.4), 3.30 (1H, dd, J=13.3, 3.1), 2.72 (1H, dd, J=13.3, 9.8), 1.94 (9H, s); $\delta_{\rm C}$ (90.6 MHz, CDCl₃) 170.7, 153.7, 139.5 (2C), 137.7, 136.2 (4C), 135.8, 133.6, 130.3, 130.2, 129.9 (2C), 129.7 (3C), 129.4 (2C), 129.0, 128.9 (3C), 128.8, 128.7, 128.6 (2C), 128.2 (4C), 128.1 (2C), 127.9, 127.8, 127.6 (2C), 77.8, 72.8, 68.2, 67.2, 62.3, 60.3, 56.2, 54.7 (2C), 38.2, 27.6 (3C), 19.6. **20**; $[\alpha]_D$ -33.6 (c 1.00, CHCl₃); v_{max} (neat)/cm⁻¹ 3451, 1779, 1705; $\delta_{\rm H}$ (360 MHz, CDCl₃) 7.77–7.75 (4H, m), 7.51–7.37 (10H, m), 7.35–7.19 (16H, m), 5.46 (1H, d, J=1.7), 4.70 (1H, ddt, J=9.2, 5.2, 3.2), 4.38 (1H, br d, J=7.7), 4.26 (1H, d, J=10.7), 4.22-4.11 (4H, m), 3.99 (1H, d, J=

- 10.7), 3.98 (2H, d, J=13.7), 3.67 (2H, d, J=13.7), 3.41 (1H, dt, J=7.7, 5.3), 3.30 (1H, dd, J=13.6, 3.2), 2.74 (1H, dd, J=13.6, 9.2), 1.10 (9H, s); $\delta_{\rm C}$ (50.3 MHz, CDCl₃) 171.7, 153.1, 140.0 (2C), 137.4, 135.7 (2C), 135.6 (2C), 135.0, 133.1, 133.0, 129.7 (2C), 129.3 (2C), 129.1 (4C), 128.8 (2C), 128.1 (2C), 128.0 (4C), 127.9 (2C), 127.7 (4C), 127.5, 127.2, 126.7 (2C), 78.3, 72.2, 72.0, 66.7, 61.5, 59.7, 54.7 (3C), 37.7, 26.7 (3C), 18.9.
- 28. **22**; $[\alpha]_D$ +40.2 (*c* 1.25, CHCl₃); ν_{max} 3305, 1689; δ_H (250 MHz, CDCl₃) 7.67–7.59 (4H, m), 7.41–7.30 (6H, m), 6.52 (1H, br s), 5.00 (1H, br s), 4.42 (1H, br d, J=2.7), 4.12–3.98 (1H, br m), 3.83 (1H, dd, J=10.8, 5.2), 3.76–3.61 (2H, m), 1.00 (9H, s); δ_C (62.9 MHz, CDCl₃) 175.4, 135.5 (2C), 135.4 (2C), 132.6, 132.4, 129.8, 129.8, 127.8 (4C), 74.9 (2C), 62.6, 55.5, 26.7 (3C), 18.9.
- For recent examples of the use of this strategy see: (a) Zhang, J.; Xiong, C.; Wang, W.; Ying, J.; Hruby, V. J. Org. Lett. 2002, 4, 4029–4032; (b) Mulzer, J.; Schülzchen, F.; Bats, J.-W. Tetrahedron 2000, 56, 4289–4298; (c) Oba, M.; Miyakawa, A.; Nishiyama, K. J. Org. Chem. 1999, 46, 9275–9278; (d) Langlois, N. Tetrahedron: Asymmetry 1998, 9, 1333–1336.
- 30. **24**; $[\alpha]_D$ -33.1 (*c* 0.75, CHCl₃); v_{max} (neat)/cm⁻¹ 1794, 1761, 1719; $\delta_{\rm H}$ (250 MHz, CDCl₃) 7.66–7.58 (4H, m), 7.45–7.30 (6H, m), 4.31 (1H, t, J=1.5), 4.13–3.96 (2H, m), 3.88-3.74 (2H, m), 1.38 (9H, s), 1.05 (9H, s), 0.88 (9H, s), 0.86 (9H, s), 0.15 (3H, s), 0.13 (3H, s), 0.12 (3H, s), 0.10 (3H, s); δ_C (62.9 MHz, CDCl₃) 172.0, 149.7, 135.4 (4C), 133.0, 132.9, 129.7 (2C), 127.7 (4C), 82.9, 78.8, 72.0, 66.7, 62.6, 27.8 (3C), 26.7 (3C), 25.7 (3C), 25.6 (3C), 19.1, 18.0, 17.8, -4.5, -4.6 (2C), -5.2. **27**; $[\alpha]_D$ -30.0 (*c* 0.2, CHCl₃); $v_{\rm max}$ (neat)/cm⁻¹ 1791, 1754, 1715; $\delta_{\rm H}$ (250 MHz, CDCl₃) 7.63-7.56 (4H, m), 7.45-7.33 (6H, m), 4.14 (1H, dd, J=3.7, 2.4), 3.90–3.85 (2H, m), 3.73 (1H, m), 2.48 (1H, dq, J=7.7, 3.7), 1.43 (9H, s), 1.24 (3H, d, J=7.7),1.04 (9H, s), 0.85 (9H, s), 0.06 (3H, s), 0.00 (3H, s); $\delta_{\rm C}$ (50.3 MHz, CDCl₃) 175.5, 149.8, 135.5 (4C), 132.8, 132.5, 129.8 (2C), 127.7 (4C), 82.8, 72.7, 66.8, 62.0, 47.8, 27.8 (3C), 26.8 (3C), 25.5 (3C), 19.1, 17.7, 14.5, -4.5, -4.7.
- 31. Nectrisine **2**; $[\alpha]_D$ +21.0 (c 0.4, H₂O) [lit. 9c $[\alpha]_D$ +21.8 (c 0.6, H₂O)]; δ_H (600 MHz, D₂O) 7.72 (1H, d, J=2.4), 4.75 (1H, dt, J=5.2, 1.0), 4.11 (1H, t, J=5.3), 3.91 (1H, dd, J=11.5, 3.9), 3.88 (1H, ddddd, J=5.3, 4.2, 3.9, 2.3, 1.2), 3.79 (1H, dd, J=11.5, 4.2); δ_C (62.9 MHz, D₂O) 170.5, 83.4, 78.2, 76.8, 61.2; m/z (FAB) 131 ([M]+, 38%), 91 (100); HRMS (FAB) (C₅H₉NO₃ requires 131.0582, found 131.0582).